Compact XAFS Spectrometer

• Background

Current XAFS instruments include an ion chamber, scintillation detector, germanium detector, and silicon draft detector. However, they all have low energy resolution or none. For some measurements, such as in complicated combined samples requiring extra low background measurements, high energy resolution greater than 60eV is needed. The crystal analyzer based spectrometer is the only choice that can fulfill these requirements today.

Detector	Energy resolution
Ion chamber	None
Scintillation	keV range
Germanium	300 eV
Si Draft	130 eV

1. Principle

The spectrometer is based on a spherically bent single crystal analyzer which can diffract x-ray energy at matching Bragg angles. Each Bragg angle refers to each energy with a certain bandwidth (also called energy resolution of the analyzer). When we tune the Bragg angle to the energy of the target fluorescence line, the detector, or x-ray counter, can only receive the target fluorescence signal. If we fix the Bragg angle of the spectrometer and scan the incident x-ray beam energy, we can get XAFS spectrum in the fluorescent model.

Normally, the sample, crystal, and detector are set on the Rowland circle to get the best energy resolution, where the crystal bent radius is equal to the diameter of this Rowland circle. Sample



2.mini-spectrometer

The spectrometer consists of a crystal analyzer, a detector and attached motion stages/rotators, which can adjust their positions to fit in the Rowland circle and expected Bragg angles. The sample holder is not included since it mostly sits on a beamline. The crystal is 100mm in diameter, and bent to 182mm radius. Its entry solid angle is 2% of 4π . The spectrometer's physical size is about 300mm x 300mm x 300mm.



3.spectrometer properties

- The most important property of the spectrometer is its energy resolution, or its crystal analyzer energy resolution. It varies with Bragg angles due to bending distortion and geometry distortion.
- There is a relation between x-ray energy resolution and Bragg angle.
 - $-\Delta E = E\Delta \Theta / tan \Theta$

The larger the Bragg angle, the better the energy resolution.

For each target element, there is a different fluorescence energy. One needs to choose a suitable orientation crystal to match its Bragg law in order to maximize the Bragg angle and get better energy resolution.

We normally run an elastic scan on a synchrotron beamline to measure spectrometer energy resolution.

• For example, if we want to measure Mn K-edge XAFS, we need to monitor the Mn Kalpha line with this spectrometer. Its energy is around 5900eV, the suitable Si crystal orientation is Si(422), which gives the largest Bragg angle 71.4 deg. (currently, we can only bend Si). This bent crystal analyzer can give an energy resolution about 18eV. At this energy, it can cover both Mn K-alpha1 and K-alpha2.

4. Energy resolution of the mini-spectrometer

• Si(422) elastic line at E=5900eV • Si(553) elastic line at E=8904eV



X-ray energy that fall in these ranges can be diffracted to the detector. So energy resolution ΔE , can also be regarded as energy through bandwidth or energy window for this spectrometer. Typical energy resolution for most of the ten third row transit element K-edges is around 4eV to 40eV.

5.XAFS measurement

• Mn K-edge of MnF_2 by Si(422), signal count rate at 200k/sec above edge with x-ray beam intensity at 10^{10} photons/sec



Mn-K edge of powder Pr_{0.7}Ca_{0.3}MnO₃

• Measured with minispectrometer with Si(422) • Measured with Lytle ion chamber



Note there is low energy side peak at right panel from Pr L edge

• Conclusions

- Mini-spectrometer can offer 4eV to 40eV energy resolution or energy window for most third row transit elements. Its great energy resolution can maximally reduce and eliminate background from the environment.
- If analyzers can separate K-beta main peak and its satellite peak(normally 16eV lower than main peak), one can then measure spin selected XAS(SSXAS) since these two peaks come from different spin states.(ref: V60, 4665, Physical Review B, 1999).
- The spectrometer has a big entry solid angle due to its small bent radius and a big size in order to get the most fluorescence signals from samples. It is better suited for the third generation synchrotron facility due to its much higher intensity beam.
- The spectrometer's compact size allows it to be mounted easily and combined with other instruments, such as the Hube diffractometer.